

# Human Molecular Genetics 2nd Edition

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web this data shows that there was a gain the mass of the lighter which shows that there was a certain amount of butane 0 38 grams 0 02 grams gas that was successfully collected during the gas collection there is a mild influence of barometric pressure during both trials as shown above in the raw data

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web 11 use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 4 l at stp v 1 n 1 v 2 n 2 0 0896 l n 22 4 l 1 mole n 0 0040 moles 12 experimental molar mass g mole of butane 0 23 g 0 0040 moles 58 g mole 13 the molecular formula of butane is c 4 h 10

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web nov 3 2020 in this experiment the moles and mass will be required to determine the molar mass of butane in

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58 g mol

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displacement method

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web in this lab you will collect a  
measurable volume of butane gas using  
water displacement using dalton s law  
and the ideal gas law the number of  
moles of butane gas and then the molar  
mass of butane can be calculated your  
experimental value for molar mass will  
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that butane is an ideal gas and that one  
mole has a volume of 22 l at stp  
experimental molar mass g mole of  
butane the molecular formula of butane  
is c 4 h 10

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web although the most convenient gas  
for this is butane other gases may be  
available in similar small cylinders the  
collection of a gas sample and the  
weighing of the gas cylinder before and  
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winterbourne cannot see his aunt  
because she is not well she has a  
headache he is an american boy about  
nine or ten years old he has a pale face  
he is wearing short trousers with long  
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